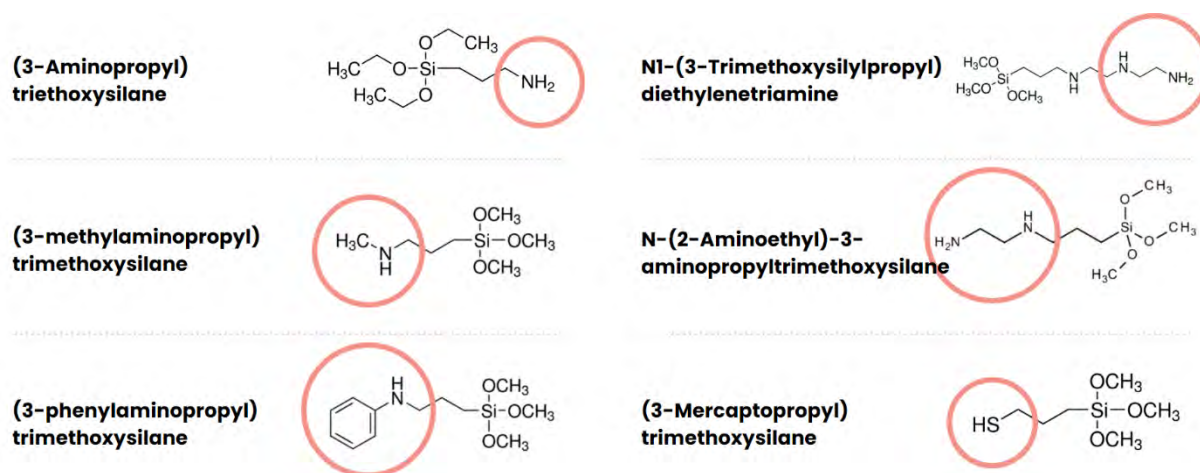


Fe₃O₄@SiO₂ Core-Shell Material with Different Ligands on SurfaceN. Király^{a*}, E. Beňová^a, V. Zelenák^a, A. Zelenáková^b^aDepartment of Inorganic Chemistry, Institute of Chemistry, Faculty of Science, Pavol Jozef Šafárik University in Košice, Moyzesova 11, 040 01 Košice, Slovak Republic^bDepartment of Condensate Matter Physics, Institute of Physics, Faculty of Science, Pavol Jozef Šafárik University in Košice, Park Angelinum 9, 041 54 Kosice, Slovak Republic

*nikolas.kiraly@upjs.sk

Thus, so far, magnetic nanoparticles especially Fe₃O₄ become the study focus of material scientists due to their unique physicochemical and high application potentials. Silica has more stable properties towards acidic conditions, has hydroxyl groups that enable the functionalization of Fe₃O₄ to bind diverse biological ligands. Another excellence related to the properties of SiO₂ is having a nanometre particle size with interface energy and adequate bounds to connect with the core. Besides, SiO₂ particles in nontoxic, biocompatible state, also have hydrophilic properties due to the existence of a silanol group on the surface. The combination of Fe₃O₄ and SiO₂ in the nanocomposite system has more advantages such as having biocompatibility, high biostability, and excellent response in drug delivery [1-3]. In recent work, we present Bio-On-Magnetic-Beads (BOMB) [4] synthetic route. We successfully prepare nano-Fe₃O₄ cores using precipitation reaction. In the next step of the work, we covered cores with a non-porous silica shell. The resulting core-shell material were be modified with several selected ligands with amine or mercapto groups (Figure 1). Modified core-shell materials were fully characterized using elemental analysis, Fourier-transform infrared (FTIR) spectroscopy, thermogravimetric analysis and adsorption techniques. The elemental analysis confirmed containing functional ligand on the surface of core-shell material. The thermal analysis was characterised in the range of 50 to 900 °C and confirmed the presence of different ligands on the surface of the core-shell materials. The infrared spectrum of Fe₃O₄@SiO₂ nanocomposites was characterized using FTIR spectroscopy which was run at the wavenumbers ranging between 4000-400 cm⁻¹. This analysis was used to study the organic and inorganic bonds formed in Fe₃O₄@SiO₂ nanocomposites.

**Figure 1** List of used ligands for modification**Acknowledgements**

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